

# Depolymerization of Microbial-Derived Polyhydroxybutyrate (PHB) in an Aqueous Environment

Re-Inventing the Nation's Urban Water Infrastructure (ReNUWIt)



Morgan Sulzbach<sup>1</sup>, Shijie Leow<sup>2,3,4</sup>, Timothy Strathmann<sup>3</sup>

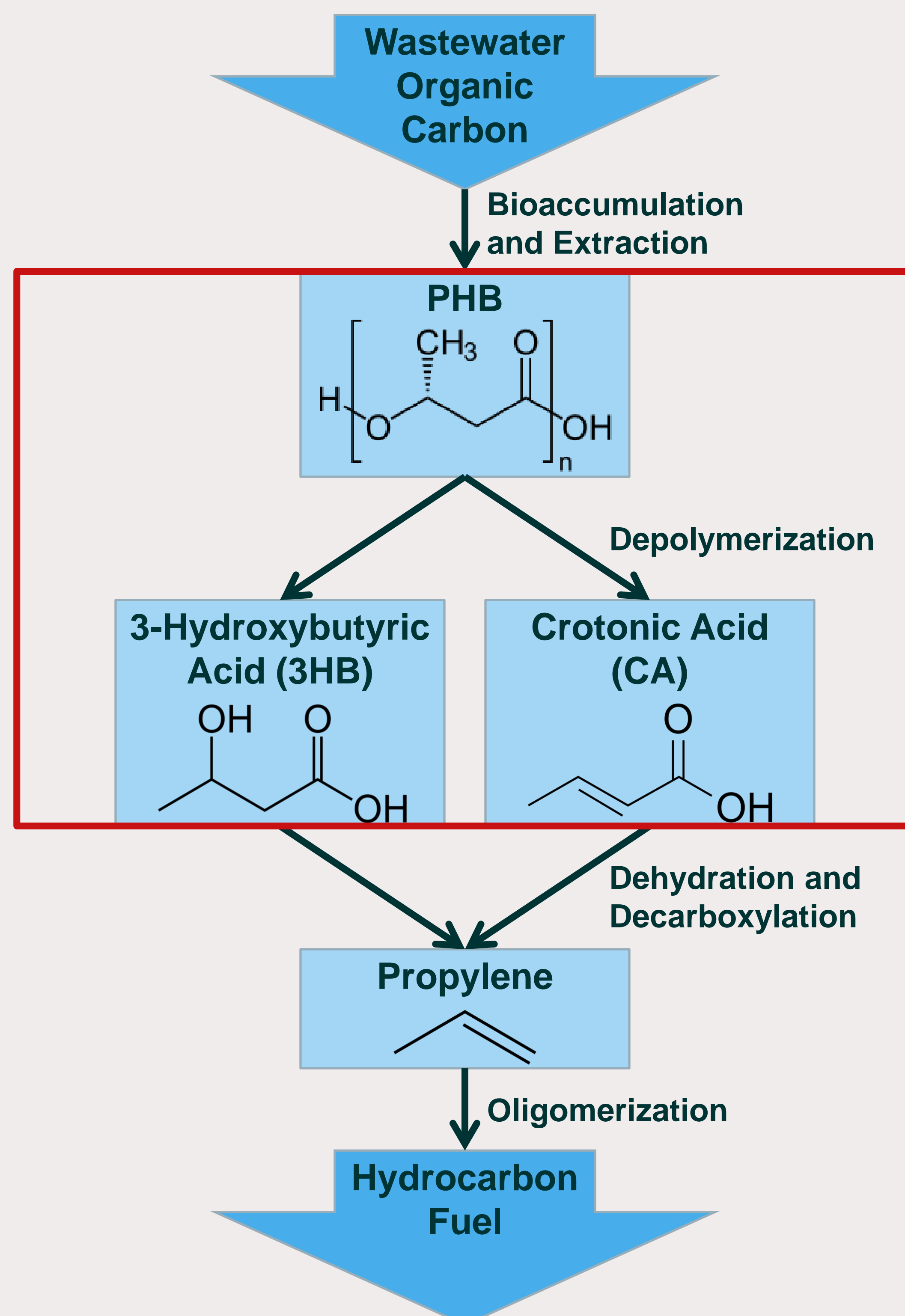
<sup>1</sup>University of Maryland, College Park, <sup>2</sup>University of Illinois at Urbana-Champaign,

<sup>3</sup>Colorado School of Mines, <sup>4</sup>National Renewable Energy Laboratory

## Introduction

**Problem:** Current processes to obtain PHB-rich biomass focus on high-quality plastic extraction which leads to prohibitively high costs.

**Solution:** An alternative approach is to accumulate PHB for carbon storage to form a biofuel from wastewater.



**Figure 1.** Overall process scheme for upgrading PHB derived from wastewater organic carbon into a hydrocarbon fuel.

## Methods



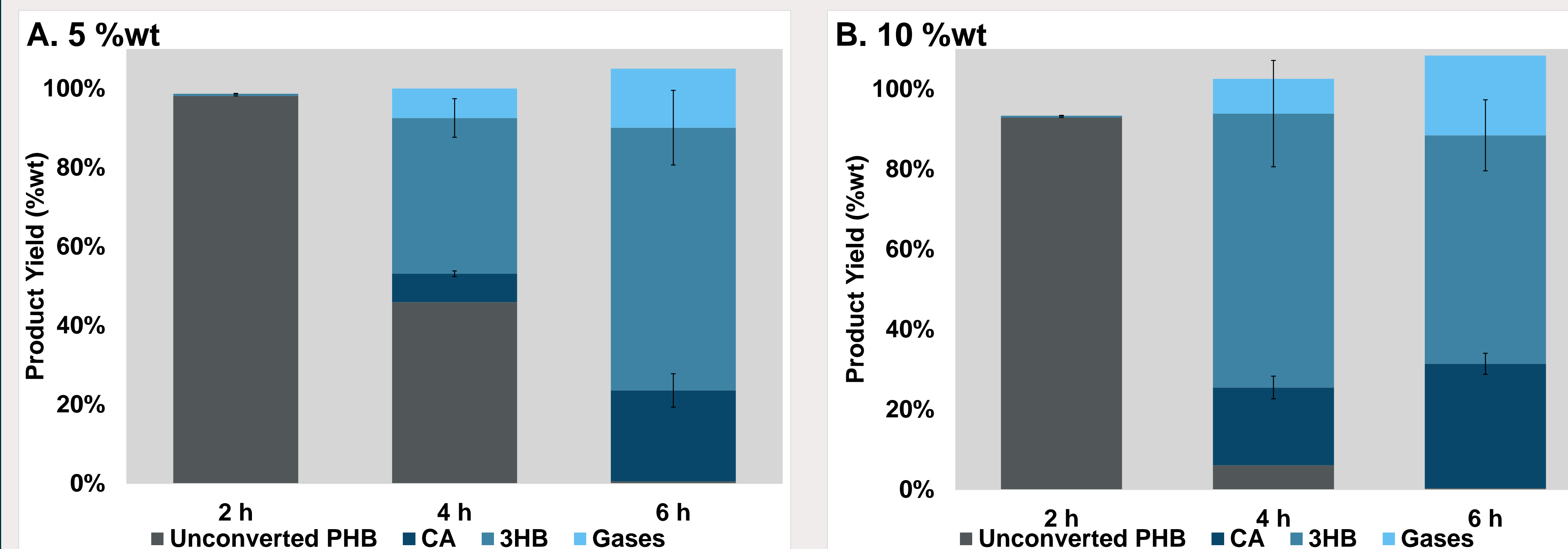
**Figure 2.** Aqueous slurries (5 or 10 %wt PHB) were reacted in 3" steel tubes placed inside a furnace. Liquid products were analyzed using HPLC.

## Results

### Objective:

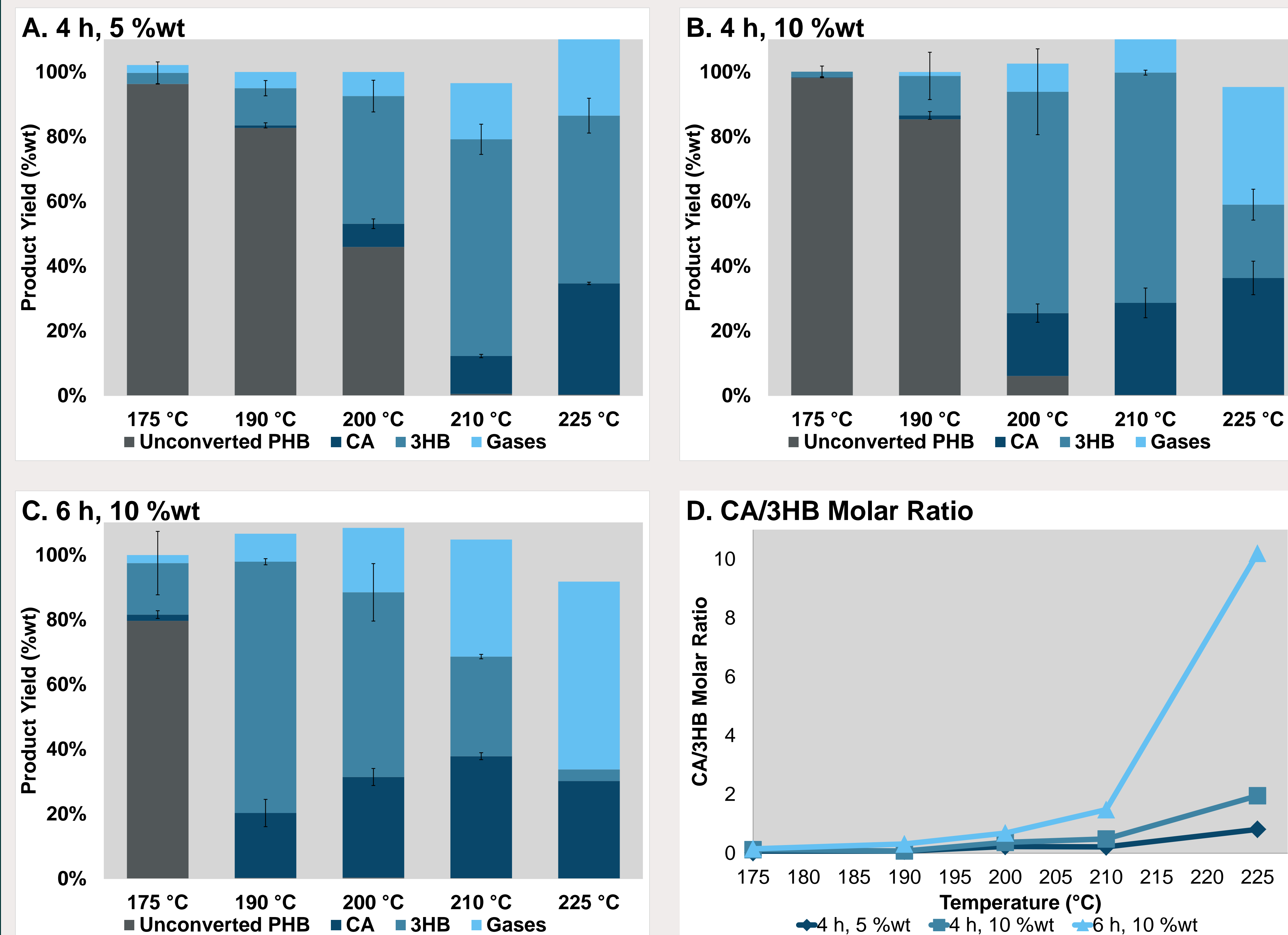
- Depolymerize PHB to CA and 3HB at high yields while minimizing energy demands.

### Time and Loading Effects



**Figure 3.** Product yields as a function of time at 200°C and A. 5 %wt and B. 10 %wt. Unrecovered yields were counted towards unconverted PHB.

### Temperature Effect



**Figure 4.** Product yields as a function of temperature at A. 4 h, 5 %wt, B. 4 h, 10 %wt, and C. 6 h, 10 %wt alongside D. the molar ratio of CA to 3HB. Unrecovered yields were counted towards unconverted PHB.

## Observations

- Higher temperatures and longer times lead to a higher CA/3HB ratio which is hypothesized to make downstream processing easier.
- The same conditions result in more gas production; however, propylene may be one of the gases produced.

## Conclusions

- PHB depolymerization in aqueous conditions and the resulting CA/3HB monomer ratio can be successfully controlled by the reaction conditions.
- Results support the final goal of having an integrated process to convert PHB to propylene all under aqueous conditions.

## Future Directions

- Replicate results in a batch reactor with stirring.
- Test at lower temperatures under acidic and basic conditions.
- Optimize each step and perform life cycle analysis on the entire process.

## Significance

- The process would add value to a waste stream while recycling carbon to make an energy-dense fuel.

## Acknowledgements

- ReNUWIt program and Colorado School of Mines for hosting the REU
- My mentors listed above and all members of the lab who have helped me throughout the summer
- This material is based upon work supported by the National Science Foundation under Grant No. EEC-1559984. Any opinions, findings, and conclusions or recommendations expressed in this material are those of the author(s) and do not necessarily reflect the views of the National Science Foundation.

